## **Significant Figures**

1. Why are significant figures important?

Significant figures indicate the degree of certainty that a value has. Your measurements and calculations can only be as precise as the least precise instrument. More digits carried does not mean a more correct answer, if you are using more digits than the instrument used accounts for.

- 2. How many significant figures are in the following numbers
  - a. 30,100 Three Only the 301 - the zeroes following the 1 are place holders
  - b. 30,000. Five
    All of the digits are significant as indicated by the decimal following the final zero.
  - c. 50,000 One
    Only the 5 is significant as there is no decimal. The rest of the zeroes are place holders.
  - d. 0.00030 Two
    Everything after the first non-zero integer in a decimal is significant.
  - e. 2.54 exactly Infinite There is no uncertainty in an exact number.
  - f.  $5.82 \times 10^3$  Three In scientific notation, the number preceding the x 10 contains all the significant figures.
- 3. Rules for carrying over significant figures

a. Addition/Subtraction

Based on number that has fewest decimal places.

i. How many significant figures would the answer have for 5.42 +13.159 = 18.579

As 5.42 is only certain to the hundredths rather than thousandths the answer must be reported as **18.58** 

b. Multiplication/Division

Based on the value that has the smallest number of digits.

i. How many significant figures would the answer have for (53298) x (432) = 23024736

As 432 has only 3 sig figs the answer must be reported as:

**2.30** x 10<sup>7</sup> or 230. x 10<sup>5</sup>

4. What is dimensional analysis?

Conversion from one set of units to another.

5. Perform the following calculation

a.  $32 \text{ cm}^3 = \_\_\_ \text{L}$ 

 $32 \text{ cm}^3 \frac{\text{mL}}{\text{cm}^3} \frac{L}{1000 \text{mL}} = 0.032L$ 

b. 100.0 yards = \_\_\_\_cm

$$\frac{100.0 \text{ yards}}{\text{yard}} \quad \frac{3 \text{ ft}}{\text{ft}} \quad \frac{12 \text{ ineh}}{\text{ineh}} = 9144 \text{ cm}$$

all conversions used were exact, so none of them limit the number of digits the answer must have.

c. 
$$5 \text{ m}^3 = \_\_\_ \text{ cm}^3$$

$$\frac{5 m^{3}}{m^{3}} \frac{(100)^{3} cm^{3}}{m^{3}} = 5,000,000 cm^{3}$$

d. 455 seconds = \_\_\_\_ minutes

455 seconds <u>1 minute</u> = 7.58 minutes 60 seconds

e. 25.6°C = \_\_\_\_ K

25.6°C + 273.15 = 298.7 K